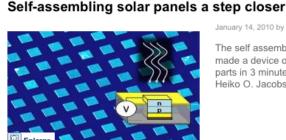
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Biochemistry Polymers Analytical Chemistry **Materials Science**



cells self-assemble.

January 14, 2010 by Lin Edwards The self assembly process made a device of 64,000 parts in 3 minutes. Image:

Heiko O. Jacobs

(PhysOrg.com) -- Scientists Robert J. Knuesel and Heiko O. Jacobs of

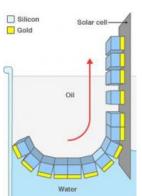
The researchers had previously been unsuccessful in their attempts to make self-assembling electronic components. In large systems gravity can be used to drive self-assembly, and in nanoscale systems chemical processes can be used, but between the two scales, in the micrometer range, it is much more difficult.

the University of Minnesota have developed a way to make tiny solar

To overcome the difficulties, Kneusel and Jacobs designed a flexible substrate of a thin layer of copper covered with propylene-terephthalate (PET). Regular depressions the same size as the "chiplets" were etched into the PET layer and then the sheet was dipped into a bath of molten solder, which coated the exposed copper in the etched depressions. Each chiplet consisted of a 20-60 µm silicon cube with one gold face. The silicon sides had a coating of hydrophobic (water-repelling) molecules, while the gold side had a hydrophilic (water-attracting) coating.

When the elements were placed in a container containing oil and water, they neatly arranged themselves in a sheet at the boundary between the liquids, with the gold side pointed down to the water layer. The substrate was then pulled slowly up through the boundary like a conveyor belt, and the elements neatly dropped in place in the depressions as the solder attracted the gold side. Accuracy was 98%. The assembly was covered with epoxy to keep the chiplets in place, and then a conducting electrode layer was added.

The device was able to assemble 62,000 elements, each of them thinner than a human hair, in only three minutes. The elimination of a dependency on gravity and sedimentation meant the chiplets could be reduced to below 100 micrometers in size. It was important to limit the assembly time to avoid oxidation of the surfaces, which would reduce surface energies and interfere with self-assembly. The water layer had to be acidic, at pH 2.0, and the temperature had to be kept at 95C to keep the solder molten.



Enlarge

The self assembly process. Elements align at the oil/water boundary. The "blank" solar cell has pre-cut places for the elements and is dipped through the boundary. As it is slowly drawn upwards, the elements pop into place. Image: Heiko O. Jacobs.

The researchers think they can adapt their method to smaller components and larger assembled devices, and it could be used to cheaply and quickly assemble all kinds of high-quality electronic components on a wide range of flexible or inflexible substrates including plastics, semiconductors and metals. The assemblages could find uses in numerous applications such as solar cells, video displays and tiny semiconductors.

The use of this method in solar cell production would reduce the cost considerably since less silicon is needed, and it should also be possible to assemble solar chiplets into transparent, flexible materials, which would extend their range of uses.

The paper is published in the Proceedings of the National Academy of

More information: Self-assembly of microscopic chiplets at a liquidliquid-solid interface forming a flexible segmented monocrystalline solar cell, Robert J. Knuesel and Heiko O. Jacobs, PNAS, DOI:10.1073/pnas.0909482107

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And the efficiency of the resultant cell was wha?

c'mon!!!

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The linked abstract says:

As an example we demonstrate a fault-tolerant segmented flexible monocrystalline silicon solar cell, reducing the amount of Si that is used when compared to conventional

This seems a bit ambiguous, but _could_ mean that the produced solar cells are of comparable efficiency. Unfortunately they want cash for the full article.

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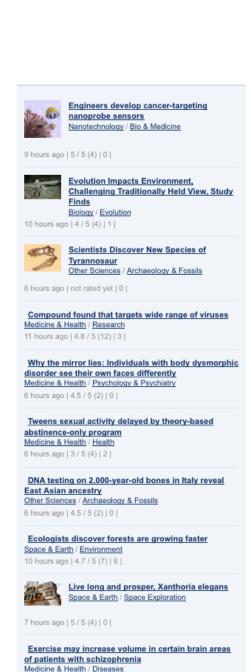
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